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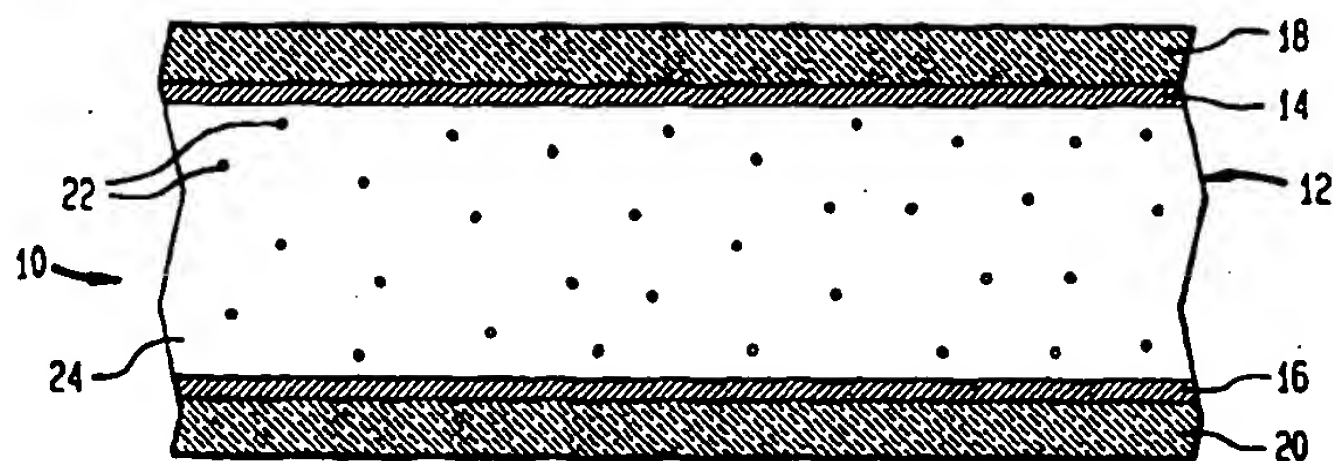
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(54) Title: BLACK AND WHITE ELECTROPHORETIC PARTICLES AND METHOD OF MANUFACTURE



## (57) Abstract

A process for forming dielectric particles (22) includes admixing a first monomer and a crosslinker in a liquid dispersion medium to form a first mixture. A second mixture of an initiator and a stabilizer is prepared and added to the first mixture to form a third mixture in which the first monomer polymerizes to form polymer particles. A second monomer is introduced to the third mixture, the second monomer at least partially polymerizing and grafting upon the previously formed polymer particles. A functional monomer effecting the outer surface charge characteristics of the final particles may be introduced to the third mixture when the second monomer is introduced. The particles (22) may be employed in an electrophoretic fluid (12) for use in an electrophoretic display by dispersing the dielectric particles (22) prepared by the two stage polymerization process in a dielectric fluid (24).

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## BLACK AND WHITE ELECTROPHORETIC PARTICLES AND METHOD OF MANUFACTURE

### TECHNICAL FIELD OF THE INVENTION

5 The present invention relates to dielectric particles and methods for producing same and more particularly to dielectric black and dielectric white particles for use in electrophoretic image displays, electrostatic printing or the like.

### BACKGROUND ART

10 The electrophoretic effect is well known and the prior art is replete with a number of patents and articles which describe the effect. As will be recognized by a person skilled in the art, the electrophoretic effect operates on the principle that certain particles, when suspended in a medium, can be electrically charged and thereby caused to migrate through the medium to an electrode of opposite charge. Electrophoretic image displays (EPIDs) utilize the  
15 electrophoretic effect to produce desired images. In prior art EPIDs, colored dielectric particles are suspended in a fluid medium that is either clear or an optically contrasting color as compared to the dielectric particles. The colored electrophoretic particles are then caused to selectively migrate to, and impinge upon, a transparent screen, thereby displacing the fluid medium against the screen  
20 and creating the desired image.

As will be recognized by a person skilled in the art, the selection of the electrophoretic particles used in the EPID is very important in determining the performance of the EPID and the quality of the viewed image produced. Ideally, electrophoretic particles should all be of a uniform size, to help in  
25 assuring that each of the electrophoretic particles will behave similarly. Additionally, it is desirable to utilize electrophoretic particles that have essentially the same density as the fluid medium in which they are suspended. By using electrophoretic particles of essentially the same density as the suspension medium, the migration of the electrophoretic particles through the medium

remains independent of both the orientation of the EPID and the forces of gravity.

5 To effect the greatest optical contrast between electrophoretic particles and the suspension medium, it is desirable to have either white particles suspended in a black medium or black particles suspended in a backlighted clear medium. In the prior art, it has been proven difficult to produce black electrophoretic particles that are dielectric, of uniform size and have a density matching that of a common suspension medium. As a result, EPIDs commonly use readily manufactured light colored electrophoretic particles suspended in dark media. Such EPIDs are exemplified in U.S. Patent Nos: 4,655,897 to DiSanto et al., 4,093,534 to Carter et al., 4,298,448 to Muller et al., and 4,285,801 to Chaing. In such prior art, light colored particles are commonly inorganic pigments which have fairly high densities. With the electric field applied, the light colored particles migrate through the grayish suspension producing a light image on a gray background, thereby resulting in an image that is not highly contrasted.

15 Although titanium dioxide used in EPIDs produces a good optical contrast between the white particles and the suspension medium, it has a density about 4 g/cm<sup>3</sup> which is too high to be matched with an organic solvent. Sedimentation becomes a problem. In the past decade, great effort has been spent to solve the density problem of titanium dioxide. Coating titanium dioxide particles with a polymeric material to reduce the density of titanium dioxide is an example. Phase separation, direct emulsification, emulsion polymerization and miniemulsion polymerization are the most common techniques used to make the polymer-coated titanium dioxide particles. The uniformity of the coating thickness and the size of final particles made by these methods cannot be well controlled, this causes a large difference in density among final particles, and balancing the density between the final particles and the suspension medium is still a difficult problem.

25  
30 In application to EPID displays, the properties of the white particles is highly specialized. First, the density of the particles must be low and

uniform in order to be balanced with the suspension medium to prevent sedimentation of the particles. Secondly, the particles must have well controlled surface functionalities for particle charging in dielectric media in order to reach an optimum electrophoretic mobility for electrophoretic imaging. Thirdly, the particles must have suitable crosslinking density as well as particle size and size distribution in order to form good whiteness yielding better optical contrast with the dark medium. Finally, the particles must have good heat and solvent resistance. Conventional techniques of making crosslinked polymer particles are suspension polymerization, emulsion polymerization, miniemulsion polymerization. Unfortunately, the properties of crosslinked polymer particles required in EPID are difficult to obtain by the conventional techniques.

Crosslinked polymer particles prepared by a suspension polymerization technique have wide particle size distribution, e.g. 1-50  $\mu\text{m}$ , which requires classification of the polymer particles. Polymer particles with a narrow particle size distribution can only be obtained at a very low yield. Emulsion polymerization can produce crosslinked polymer particles with a very narrow size distribution, however, it only can produce particle sizes in a sub-micron range (J. Appl. Phys., 26(7), 864 (1955)). In addition, only small amounts of crosslinking monomers can be used, producing particles with poor heat and solvent resistance and poor whiteness. Using the seeded emulsion polymerization technique, crosslinked polymer particles with uniform particle sizes greater than 1  $\mu\text{m}$  can be produced, however, it takes a long time to complete the whole process (Polym. Mater. Sci. Eng. 54, 587 (1986)). On the other hand, miniemulsion polymerization produces (J. Polym. Sci., Polym. Chem. Ed., 17,3069 (1979)) polymer particles having higher crosslinking density, however, the particle size distribution is too broad to obtain uniform electrophoretic mobility resulting in poor electrophoretic images.

It has been reported that it is difficult to produce stable uniform crosslinked polymer particles by a dispersion polymerization method when the

crosslinking monomers is over 1% by weight (J. Polym. Sci., Polym. Chem. Ed., 24, 2995 (1986). Reports are also found for preparing styrene/divinylbenzene particles by batch dispersion co-polymerization and seeded dispersion copolymerization respectively (Colloid Polym. Sci., 269, 217 (1991), however, good monodispersity and heat and solvent resistance still can not be obtained. More recently, Kobayashi and Senna reported production of uniform styrene/divinylbenzene polymer particles with high crosslink density using a dispersion polymerization technique (J. Appl. Polym. Sci., 46,27 (1992)). Although they claimed that the highly crosslinked polymer particles are uniform in size and are greater than 1  $\mu\text{m}$ , no claim was made in regard to controlling of surface functionalities of the final particles, which is so important for particle charging in dielectric media, particularly in EPIDs.

In addition to using white particles suspended in a dark medium to produce contrast images, one may alternatively suspend black particles in a backlighted clear medium. However, as has been mentioned, the development of suitable dielectric black particles remains a goal in the art of electrophoretic image displays. In art other than EPIDs, black particles are commonly produced from carbon. However, carbon blacks are not readily adaptable to EPIDs because carbon blacks are conductive and the density of carbon blacks is not readily matched to a suitable suspension medium. Research efforts have been made in an attempt to solve the density and conductivity problems of carbon blacks, however, none has succeeded without trading off some blackness in the particles created. Such efforts to produce dielectric particles from carbon blacks are exemplified in the following article Hou et al. "Pigmented-Polymer Particles With Controlled Morphologies", (Polymer Latexes, ACS Symposium Series 492, Chap. 25, p. 405, 1992).

### **DISCLOSURE OF THE INVENTION**

The problems and disadvantages associated with conventional dielectric particles and methods for making same are overcome by the present invention which includes a process for forming dielectric particles in which a first monomer is polymerized in a dispersion medium to form particles. A second monomer is then polymerized in the

same medium and grafts upon the previously formed particles. The surface functionality of the particles is controlled. The particles may be employed in an electrophoretic fluid for use in an electrophoretic display by dispersing the dielectric particles prepared by the two stage polymerization process in a dielectric fluid.

### **BRIEF DESCRIPTION OF THE DRAWINGS**

FIG. 1 is a diagrammatic, cross-sectional view of an EPID having particles in accordance with a first embodiment of the present invention.

FIG. 2 is a diagrammatic cross-sectional view of an alternative EPID construction having particles in accordance with a second embodiment of the present invention.

### **BEST MODE FOR CARRYING OUT THE INVENTION**

The present invention uses a two-stage dispersion polymerization technique to produce highly crosslinked polymer particles with whiteness sufficient for good image contrast in dark suspension media and which have a density close to 1 g/cm<sup>3</sup>. The particles can be used as white electrophoretic particles (instead of titanium dioxide) in EPIDs and have uniform size and low density and are easy to balance with the density of many organic solvents.

Particles in accordance with the present invention have specialized surface functionalities which permit control of particle charging in dielectric

media to produce an optimum electrophoretic mobility, which, with controlled particle size, size distribution and crosslinking density, permit production of high quality electrophoretic images.

5 The present invention may also be employed to produce black electrophoretic particles which do not use carbon black. Rather, highly crosslinked polymer particles stained with a metal oxide are used to form dielectric black particles suitable for use in an EPID. More particularly, the preferred embodiment of the present invention uses a two-stage dispersion polymerization technique to produce highly crosslinked polymer particles which  
10 are further reacted with a metal oxide to form dielectric black particles with good blackness. These particles yield good image contrast in a backlighted clear medium, have uniform size and low density which make them easy to density match with any organic solvents, and have specialized surface functionalities which allow control of particle charging in dielectric media to produce optimum  
15 electrophoretic mobility.

Although the present invention black and white particles can be used in many different applications where particles of high blackness or whiteness and low density are desired, such as coating, printing ink and electrostatic printing, they are especially suitable for use in connection with  
20 electrophoretic image displays (EPIDs). Accordingly, the present invention black and white dielectric particles will be described in connection with typical EPIDs.

Referring to Fig. 1, there is shown a cross sectional view of a segment of a simple electrophoretic image display 10, magnified to show a single pixel (intersection). As will be recognized by a person skilled in the art, an EPID  
25 may contain a volume of an electrophoretic dispersion 12 disposed between an anode 14 and a cathode 16. The anode 14 and cathode 16 are deposited upon glass plates 18, 20 in the form of a thin layer of indium-tin-oxide (ITO) or a like compound. The ITO layer is deposited in such a manner as to be substantially transparent when viewed through the glass plates 16, 18.



In accordance with a first embodiment of the present invention, the electrophoretic dispersion 12 is comprised of white dielectric electrophoretic particles 22 suspended in a dark color medium 24. The electrophoretic particles 22 have a density substantially equivalent to that of the fluid medium 24 so as to remain randomly disperse in the fluid medium 24, unaffected by the orientation of the EPID or the effects of gravity. When a sufficient electrical bias is applied between the anode 14 and cathode 16, the electrophoretic particles 22 migrate in response thereto to either the cathode 16 or anode 14 depending on polarity, and displace the dark color medium 24 adjacent to the ITO layer, thereby creating a white pixel. Reversing the voltage produces a dark pixel.

Figure 2 shows in cross section a pixel of another type of EPID 110 wherein electrophoretic dielectric particles are employed in the manner of a shutter. As will be recognized by a person skilled in the art, an EPID of this type contains a volume of an electrophoretic dispersion 112 disposed between an anode 114 and a cathode 116 as in FIG. 1. A conductive mesh 126 is disposed in the fluid 112 between the cathode 116 and anode 114.

In accordance with a second embodiment of the present invention, the electrophoretic dispersion 112 is comprised of black dielectric electrophoretic particles 122 suspended in clear medium 124. The electrophoretic particles 122 have a density substantially equivalent to that of the fluid medium 124 so as to remain randomly disperse in the fluid medium 124, unaffected by the orientation of the EPID or the effects of gravity. When the cathode 116, anode 114 and the mesh 126 are properly biased, the electrophoretic particles 122 may be made to migrate to cathode 116 displacing the clear medium 124 adjacent to the ITO layer, thereby blocking the light produced by the light source 128. Alternately, the electrodes may be biased so that the particles 122 cling to the mesh and do not block light from source 128. The displayed pixel is therefore black or white depending on the voltage settings.

As noted previously, the production of a white image on a dark

color background or a black image on a light background is highly desirable. However, a major obstacle to such a combination has been the lack of dielectric black and white particles that have good hiding power and also a density that can be readily matched with common suspension fluids. The present invention black electrophoretic particles 122 and white electrophoretic particles 22 are formed from crosslinked polymer particles using two stage dispersion polymerization techniques with and without staining with a metal oxide respectively. In general, the polymeric materials are dielectric and have densities close to 1 g/cm<sup>3</sup>, thus the final electrophoretic particles are non-conductive and can easily be matched with many common EPID suspension fluids without any sedimentation problem. Since the electrophoretic particles are highly crosslinked polymer particles, they have excellent heat and solvent resistance and can be used in a wide range of organic solvents, even at high temperatures during cell sampling. In addition, the particle size, size distribution and surface functionalities of the final particles can be precisely controlled during the second-stage polymerization to produce electrophoretic particles with optimum electrophoretic mobilities in dielectric media for developing high quality images.

The dielectric white electrophoretic particles 22 of highly crosslinked polymer with controlled surface functionality are prepared by a two-stage dispersion polymerization technique. As will be recognized by a person skilled in the art, two-stage dispersion polymerization involves dispersing vinyl monomers and a crosslinker in an organic solvent in which the monomers and crosslinker are soluble and then polymerizing the vinyl monomers and the crosslinker in the presence of a dispersion stabilizer and an initiator at elevated temperatures. The polymerized monomer is insoluble in the solvent, thereby forming polymer particles. The second stage functional monomer is copolymerized on the insoluble polymer particles by injecting the functional monomer at a later stage of polymerization.

EXAMPLE ONE

In an exemplary embodiment of the present invention, dielectric, white, electrophoretic, highly crosslinked polymer particles 22 with controlled surface functionality, are prepared by a two-stage dispersion polymerization technique. The polymerization recipe is listed in Table I.

Table I

<u>Materials</u>	<u>Weight (g)</u>
Methanol	100
divinylbenzene	5
styrene	5
2,2'-azobisisobutyronitrile	0.5
poly(vinyl pyrrolidone)	2
Acrylamide	0.5

Prior to use, the inhibitors of the monomer styrene and the crosslinker divinylbenzene are removed by washing with 10% NaOH aqueous solution several times, drying with calcium carbonate overnight at 0°C and then passing through a column containing an appropriate inhibitor remover. The inhibitor-free monomers and crosslinker are stored in a freezer for later use. In this and subsequent processes, the styrene is of the type commercially available from Fisher Scientific, Inc. and the divinylbenzene is of the type commercially produced by Aldrich Chemical, Co. The 2,2'-azobisisobutyronitrile (AIBN) and poly(vinyl pyrrolidone) (PVP) used as an initiator and stabilizer are manufactured by Kodak and GAF Co. respectively. The second stage monomer, acrylamide, and the dispersion medium, methanol, a reagent grade, are of the types commercially available from Fisher Scientific, Inc.

The styrene and divinylbenzene are mixed with methanol and charged to a closed container containing the AIBN and PVP which are carefully weighed. The closed container is purged with nitrogen by bubbling the gas through the solution for a certain time. The container is then warmed and

agitated for a desired reaction time. In one preferred embodiment, the mixture is tumbled at thirty revolutions per minute for eight hours at sixty degrees celsius. After eight hours tumbling, the second stage monomer, acrylamide, is injected into the container which continues to tumble at the same reaction condition for another desired reaction time. The final product made by the two-stage dispersion polymerization process is highly crosslinked poly(styrene-co-divinylbenzene) particles with polyacrylamide grafted on the surface. The particle size of the final particles are uniform and varied from 0.2 to 10  $\mu\text{m}$  depending upon the reaction media be used. The final particles are dielectric with good whiteness and have a density close to 1  $\text{g/cm}^3$ . The polymer particles are separated from the dispersion medium by centrifuging and decanting the dispersion medium.

To form black particles, the product of the previous process is mixed and tumbled with osmium tetroxide aqueous solution at room temperature for a desired reaction time. The osmium tetroxide reacts with and stains the residual double bonds of the poly(styrene-co-divinylbenzene) particles, thereby resulting in highly crosslinked polymer particles having a desired degree of blackness that can be used as the present invention electrophoretic particles 122. It should be understood that in place and stead of the osmium tetroxide, ruthenium tetroxide or other metal oxides may also be used.

By varying the polymerization recipe of Table I and by varying other reaction parameters of the method of manufacture, the physical characteristics of the white and black particles produced can be selectively altered as needed for a given application. The surface functionality of the final particles can be varied by introducing different functional monomers, such as vinyl acetate, methyl methacrylate, acrylonitrile, acrylamide, dimethylaminopropylmethacrylamide, and the like, at the second stage polymerization to produce poly(styrene-co-divinylbenzene) particles with basic surface characteristics which are suitable for positive charging in dielectric media.

The use of a dispersion stabilizer in the preparation of crosslinked polymer particles is essential for preventing particle coagulation during polymerization. Various synthetic high molecular weight compounds including homopolymers, copolymers, graft polymers, block copolymers, and the like, can be used as a dispersion stabilizer. In addition to poly(vinyl pyrrolidone) other suitable dispersion stabilizers are polyisobutylene succinimide, poly(methyl-methacrylate)-g-poly(12-hydroxy stearic acid) and the like.

The monomer, styrene, can be substituted by different vinyl monomers such as methyl methacrylate, vinyl acetate, acrylate, ethyl vinylbenzene, vinylpyridine, acrylonitrile, and the like, to produce final particles with different properties such as density, reflective index, heat resistance, solvent resistance, . . . etc. The crosslinker, divinylbenzene, can also be substituted for with other compounds having two or more than two polymerizable double bonds such as nonconjugated divinyl compounds, diacrylate compounds, triacrylate compounds, dimethacrylate compounds, trimethacrylate compounds, and the like.

Other radical polymerization initiators which can be used in the two stage dispersion polymerization are 4,4'-azobis(r-cyanopentanic acid), 2,2'-azobis(2-methylbutyronitrile), benzoyl peroxide, lauroyl peroxide, octanoyl peroxide, and the like.

The dispersion medium, methanol, can be substituted for with other alcohols such as ethyl alcohol, propyl alcohol, butyl alcohol, amyl alcohol, benzyl alcohol, ether alcohol, and the like, or hydrocarbons such as hexane, octane, decane, cyclohexane, xylene, and the like, or with ethers, halogenated hydrocarbons, ketones, esters and the like.

In addition, the ratio of the monomer and crosslinker in the Table I can be varied to determine the physical properties of the final particles, such as density, reflective index, heat resistance, solvent resistance, particle size and so on.

After the second stage polymerization process, the final polymer

particles are transferred from the dispersion medium to a desired dielectric medium by a washing process which comprises mixing the final product with a solvent which is mixable with both the dispersion medium and the desired dielectric medium, centrifuging the mixture to separate the particles and the liquids and decanting the supernatant to yield the concentrated particles. The washing procedure is repeated several times until the methanol is completely removed. The solvent washed particles are then dispersed in the desired dielectric medium with the addition of a charge control agent and a dark color dye. Sometimes, a stabilizer or a co-stabilizer is added to the final dispersion to help prevent the particles from coagulating. Therefore, the final dispersion contains electrophoretic particles having a positive electrophoretic mobility dispersed in a dark color-dye solution which gives good optical contrast with the white particles, and is ready to be used in EPIDs.

In the case of black particles, the metal oxide stained particles are transferred from the aqueous staining solution to a desired dielectric medium by the same washing process as outlined above. After being transferred to the desired medium, a charge control agent, sometimes together with a stabilizer or a costabilizer, is added to the dispersion to produce a final black dispersion containing positively charged black particles suspended in a clear medium, which is ready to use in EPIDs.

The preferred dielectric media for the final dispersion are non-polar solvents such as tetrachlorethylene, carbon tetrachloride, pentane, octane, decane, cyclohexane, benzene, xylene, Isopars, sec-butylbenzene, and the like, or a mixture of the liquids. The solvents used for washing the particles should be mixable with both the dispersion medium and the dielectric medium, examples are ethanol, propanol, butanol, acetone, tetrahydrofuran, ketones, ethers, esters, and the like. The desirable charge control agents for positive charging are surfactants with an acidic character such as polyisobutylene succinic anhydride, cupric naphthenate, zirconium octoate, zinc octoate, calcium octoate, polyvinyl

alcohol, polyacrylic acid, polymethacrylic acid, polyvinyl chloride, and the like. A dark colored dye such as Solvent Blue, Oil Blue A, Sudan Black B, Sudan Red 7B and the like, is preferred. The preferable stabilizers or co-stabilizer used with the charge control agents are high molecular weight compounds including homopolymers, copolymers, graft polymers, block copolymers, natural high molecular compounds, and the like. Specific examples of suitable stabilizers are poly(12-hydroxystearic acid)-graft-poly(methyl methacrylate-methacrylic acid), polystyrene-co-poly(vinyl pyridine), poly(vinyl alcohol)-co-polyethylene, and the like.

#### EXAMPLE TWO

In an alternative embodiment of the present invention, white, electrophoretic, highly crosslinked polymer particles 22 with controlled surface functionality, are prepared by a two-stage dispersion polymerization technique. The polymerization recipe is listed in Table II.

Table II

<u>Materials</u>	<u>Weight (g)</u>
Methanol	100
divinylbenzene	5
styrene	5
2,2'-azobisisobutyronitrile	0.5
poly(acrylic acid)	2
methacrylic acid	0.5

As in process one, the inhibitors of the monomer styrene and the crosslinker divinylbenzene are removed by washing with 10% NaOH aqueous solution several times, drying with calcium carbonate over night at 0°C and then passing through a column containing an appropriate inhibitor remover. The inhibitor-free monomers and crosslinker are stored in a freezer for later use. In this, and subsequent processes, the styrene is of the type commercially available from

Fisher Scientific, Inc. and the divinylbenzene is of the type commercially produced by Dow Chemical, Inc. The initiator, 2,2'-azobisisobutyronitrile (AIBN) is manufactured by Kodak Co. The second stage monomer, methacrylic acid and the stabilizer, poly(acrylic acid), are of the types commercially available from Aldrich Chemical Co. The dispersion medium, methanol, a reagent grade, is from Fisher Scientific, Inc.

The styrene and divinylbenzene are mixed with methanol and charged to a closed container containing the AIBN and poly(acrylic acid) which are carefully weighed. The closed container is purged with nitrogen by bubbling through the solution for a certain time. The container is then warmed and agitated for a desired reaction time. In one preferred embodiment, the mixture is tumbled at thirty revolutions per minute for eight hours at sixty degrees celsius. After eight hours of tumbling, the second stage monomer, methacrylic acid, is injected in the container which continues to tumble at the same reaction condition for another desired reaction time. The final product made by the two stage dispersion polymerization process is highly crosslinked poly(styrene-co-divinylbenzene) particles with poly(methacrylic acid) grafted on the surface. The particle size of the final particles is uniform and varied from 0.1 to 5  $\mu\text{m}$  depending upon the reaction media used. The final particles are dielectric and have a density close to 1  $\text{g/cm}^3$ . The polymer particles are separated from the dispersion medium by centrifuging and decanting the dispersion medium.

To form black particles, the particles of the previous process are then mixed and tumbled with a osmium tetroxide aqueous solution at room temperature for a desired reaction time. The osmium tetroxide reacts with and stains the residual double bonds of the poly(styrene-co-divinylbenzene) particles, thereby resulting in highly crosslinked polymer particles having a desired degree of blackness that can be used as the present invention electrophoretic particles 122. It should be understood that in place and stead of the osmium tetroxide, ruthenium tetroxide or other metal oxides may also be used.



By varying the polymerization recipe of Table II and by varying other reaction parameters of the method of manufacture, the physical characteristics of the white and black particles produced can be selectively altered as needed for a given application. The surface functionality of the final particles can be varied by introducing different functional monomers, such as acrylic acid, sodium styrene sulfonate, maleic acid, chlorostyrene, vinyl alcohol, and the like, at the second stage polymerization to produce poly(styrene-co-divinylbenzene) particles with acidic surface characteristics which are suitable for developing negative electrophoretic mobilities in dielectric media.

The use of a dispersion stabilizer in the preparation of crosslinker polymer particles is essential for preventing particle coagulation during polymerization. Various synthetic high molecular weight compounds including homopolymers, copolymers, graft polymers, block copolymers, and the like, can be used as a dispersion stabilizer. In addition to poly(acrylic acid), other suitable dispersion stabilizers are poly(vinyl alcohol), poly(methacrylic acid), sorbitan stearic acid monoester, and the like.

The monomer, styrene, can be substituted for by different vinyl monomers such as methyl methacrylate, vinyl acetate, acrylate, ethyl vinylbenzene, vinylpyridine, acrylonitrile, and the like to produce final particles with different properties such as density, reflective index, heat resistance, solvent resistance, . . . etc. The crosslinker, divinylbenzene, can also be substituted for by other compounds having two or more than two polymerizable double bonds such as nonconjugated divinyl compounds, diacrylate compounds, triacrylate compounds, dimethacrylate compounds, trimethacrylate compounds, and the like.

Other radical polymerization initiators which can be used in the two stage dispersion polymerization are 4,4'-azobis(4-cyanopentonic acid), 2,2'-azobis(2-methylbutyronitrile), benzoyl peroxide, lauroyl peroxide, octanoyl peroxide, and the like.

The dispersion medium, methanol can be substituted for with other

alcohols such as ethyl alcohol, propyl alcohol, butyl alcohol, amyl alcohol, benzyl alcohol, ether alcohol, and the like, or with hydrocarbons such as hexane, octane, decane, cyclohexane, xylene, and the like, or with ethers, halogenated hydrocarbons, ketones, esters and the like.

5 In addition, the ratio of the monomer and crosslinker in the Table II can be varied to determine the physical properties of final particles, such as density, reflective index, heat resistance, solvent resistance, particle size and so on.

10 After the second stage polymerization process, the final polymer particles are transferred from the dispersion medium to a desired dielectric medium by a washing process which comprises mixing the final product with a solvent which is mixable with both the dispersion medium and the desired dielectric medium, centrifuging the mixture to separate the particles and the liquids and decanting the supernatant to yield the concentrated particles. The  
15 washing procedure is repeated several times until the methanol is completely removed. The solvent washed particles are then dispersed in the desired dielectric medium with the addition of a charge control agent and a dark color dye. Sometimes, a stabilizer or a co-stabilizer is added to the final dispersion to help prevent the particles from coagulating. Therefore, the final dispersion  
20 contains white electrophoretic particles having negative electrophoretic mobility dispersed in a dark color dye fluid which gives a good optical contrast with the white particles, and is ready to be used in EPIDs.

25 In the case of black particles, the metal oxide stained particles are transferred from the aqueous solution to a desired dielectric medium by the same washing process described above. After being transferred to the desired medium, a charge control agent, sometimes together with a stabilizer or a costabilizer, is added to the dispersion to produce a final black dispersion containing negatively charged black particles suspended in a clear medium, which is ready to use in EPIDs.

The preferred dielectric media for both the black and white dispersion are non-polar solvents such as tetrachloroethylene, carbon tetrachloride, pentane, octane, decane, cyclohexane, benzene, xylene, Isopars, sec-butylbenzene, and the like or a mixture of the liquids. The solvents used for washing the particles should be mixable with both the dispersion medium and the dielectric medium, examples are ethanol, propanol, butanol, acetone, tetrahydrofuran, ketones, ethers, esters, and the like. The desirable charge control agents for negative charging are surfactants with a basic character such as polyisobutylene succinimide, barium petronate, barium sulfonate, barium dinonylnaphthalenesulfonate, metal oxides, polyvinyl pyridine, pyridine, lecithin, polyvinyl acetate, polyethylene oxide, polymethy methacrylate, polyacrylamide, polyesters, polyethers, and the like. A dark colored dye such as Solvent Blue, Oil Blue A, Sudan Black B, Sudan Red 7B, and the like is preferred. The preferable stabilizers or co-stabilizer used with the charge control agents are high molecular weight compounds including homopolymers, copolymers, graft polymers, block copolymers, natural high molecular compounds, and the like. Specific examples of suitable stabilizers are poly(12-hydroxystearic acid)-graft-poly(methyl methacrylate-methacrylic acid), polystyrene-co-poly(vinyl pyridine), poly(vinyl alcohol)-co-polyethylene, and the like.

As will be recognized by a person skilled in the art, the highly crosslinked polymer particles having surfaces with controlled acidic moieties or basic moieties, made by emulsion polymerization, miniemulsion polymerization, microemulsion polymerization, suspension polymerization, precipitation, seeded emulsion polymerization or seeded dispersion polymerization, can also be used as white electrophoretic particles, or be used as black electrophoretic particles after being stained with a metal oxide in electrophoretic image displays.

All equivalents, variations and modifications that can be applied to the described present invention by a person skilled in the art, are intended to be included within the scope of this invention as defined by the appended claims.

**Claims:**

1. A process for forming dielectric particles, comprising:
  - (a) providing a dispersion medium;
  - (b) polymerizing a first monomer in said medium to form  
5 polymer particles;
  - (c) polymerizing a second monomer in said medium, said  
polymerized second monomer grafting upon said polymer particles; and
  - (d) controlling the surface functionality of said particles.
- 10 2. The process of Claim 1 wherein said medium is a liquid, wherein  
said step (b) includes
  - admixing a first monomer and a crosslinker with said liquid  
dispersion medium forming a first mixture;
  - preparing a second mixture of an initiator and a stabilizer;
  - 15 adding said first mixture to said second mixture to form a third  
mixture; and
  - allowing said first monomer to polymerize within said third  
mixture to form said polymer particles, and wherein said step (c) includes
  - introducing said second monomer to said third mixture, said second  
20 monomer at least partially polymerizing and grafting upon said polymer particles.
3. The process of Claim 2, wherein said step of controlling includes

the step of introducing a functional monomer when said second monomer is introduced.

4. The process of Claim 3, wherein said first monomer is a vinyl  
5 monomer.

5. The process of Claim 3, wherein said first monomer is selected  
from the group consisting of styrene, methyl methacrylate, vinyl acetate, acrylate,  
ethyl vinylbenzene, vinylpyridine, and acrylonitrile.

10

6. The process of Claim 5, wherein said crosslinker is a compound  
having a plurality of polymerizable double bonds.

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7. The process of Claim 6, wherein said crosslinker is selected from  
the group consisting of nonconjugated divinyl compounds, diacrylate compounds,  
triacylate compounds, dimethacrylate compounds and trimethacrylate  
compounds.

20

8. The process of Claim 7, wherein said initiator is selected from the  
group consisting of 2,2'-azobisisobutyronitrile, 4,4'-azobis(r-cyanopentanic acid),  
2,2'-azobis(2-methylbutyronitrile), benzoyl peroxide, lauroyl peroxide and  
octanoyl peroxide.

9. The process of Claim 8, wherein said stabilizer is selected from the group consisting of poly(vinyl pyrrolidone), homopolymers, copolymers, graft polymers, block polymers, poly(acrylic acid), poly(vinyl alcohol), poly(methacrylic acid) and sorbitan stearic acid monoester.

5

10. The process of Claim 9, wherein said second monomer is acrylamide.

10

11. The process of Claim 10, wherein said functional monomer is selected from the group consisting of vinyl acetate, methyl methacrylate, acrylonitrile, N-(iso-utoxymethyl) acrylamide and dimethylaminopropylmethacrylamide and produce dielectric particles with a basic surface suitable for positive charging.

15

12. The process of Claim 11, wherein said particles produced are white.

20

13. The process of Claim 12, further including the step of dying said white particles another color.

14. The process of Claim 13, wherein said dying is performed by immersing said white particles in an aqueous solution of a metal oxide.

15. The process of Claim 14, wherein said metal oxide is selected from the group comprising ruthenium tetroxide and osmium tetroxide.

16. The process of Claim 9, wherein said second monomer is methacrylic acid.

17. The process of Claim 16, wherein said functional monomer is selected from the group consisting of acrylic acid, sodium styrene sulfonate, maleic acid, chlorostyrene and vinyl alcohol, producing dielectric particles with an acidic surface suitable for negative charging.

18. The process of Claim 17, wherein said particles produced are white.

19. The process of Claim 18, further including the step of dying said white particles another color.

20. The process of Claim 19, wherein said dying is performed by immersing said white particles in an aqueous solution of a metal oxide.

21. The process of Claim 20, wherein said metal oxide is selected from the group comprising ruthenium tetroxide and osmium tetroxide.

22. The process of Claim 3, wherein said step of allowing said first monomer to polymerize includes agitating and warming said third mixture.

23. The process of Claim 3, further including the step of separating said polymer particles from said third mixture by centrifuging and decanting the supernatant liquid.

24. Dielectric particles formed by the process of Claim 3.

25. Dielectric particles formed by the process of Claim 14.

26. Dielectric particles formed by the process of Claim 16.

27. Dielectric particles formed by the process of Claim 20.

28. A process for forming an electrophoretic fluid for use in an electrophoretic display, comprising:

- (a) providing a dielectric fluid;
- (b) preparing dielectric particles by a two stage polymerization process; and
- (c) dispersing said dielectric particles in said dielectric fluid.



29. The process of Claim 28, wherein said step of preparing includes admixing a first monomer and a crosslinker in a liquid dispersion medium forming a first mixture;

preparing a second mixture of an initiator and a stabilizer;

5 adding said first mixture to said second mixture to form a third mixture;

allowing said first monomer to polymerize within said third mixture to form polymer particles;

10 introducing a second monomer to said third mixture, said second monomer at least partially polymerizing and grafting upon said polymer particles; and

introducing a functional monomer to said third mixture when said second monomer is introduced.

15 30. The process of Claim 29, further including the steps of centrifuging said third mixture to separate said dielectric particles from said dispersion medium and decanting the supernatant said dispersion medium prior to said step of dispersing said particles in said dielectric fluid.

31. The process of Claim 30, further including the steps of resolating said dielectric particles after centrifuging said third mixture with a solvent miscible with said dispersion medium and with said dielectric fluid and further centrifuging said particles and decanting said solvent prior to dispersing said particles in said dielectric fluid.

32. The process of Claim 31, further including the step of adding a charge control agent to said dielectric fluid after said step of dispersing in said dielectric fluid.

33. The process of Claim 32, wherein said charge control agent is an acidic surfactant and a positive charge control agent.

34. The process of Claim 33, wherein said positive charge control agent is selected from the group consisting of polyisobutylene succinic anhydride, cupric naphthenate, zirconium octoate, zinc octoate, calcium octoate, polyvinyl alcohol, polyacrylic acid, polymethacrylic acid and polyvinyl chloride.

35. The process of Claim 32, wherein said charge control agent is a basic surfactant and a negative charge control agent.

36. The process of Claim 35, wherein said negative charge control agent is selected from the group consisting of polyisobutylene succinimide, barium petronate, barium sulfonate, barium dinonylnaphthalensulfonate, metal oxides, polyvinyl pyridine, pyridine, lecithin, polyvinyl acetate, polyethelene oxide, polymethy methacrylate, polyacrylamide, polyesters and polyethers.

37. The process of Claim 32, further including adding a dye to said dielectric fluid when said charge control agent is added.

38. The process of Claim 37, wherein said dye is selected from the group consisting of Solvent Blue, Oil Blue A, Sudan Black B, and Sudan Red 7B.

39. The process of Claim 32, further including the step of adding a stabilizer to said dielectric fluid when said charge control agent is added.

40. The process of Claim 39, wherein said stabilizer is selected from the group consisting of homopolymers, copolymers and block copolymers.

41. The process of Claim 39, wherein said stabilizer is selected from

the group consisting of poly(12-hydroxystearic acid)-graft-poly(methyl methacrylate-methacrylic acid), polystyrene-co-poly(vinyl pyridine) and poly(vinyl alcohol)-co-polyethylene.

5           42.    The process of Claim 28, wherein said dielectric fluid is a non-polar solvent.

          43.    The process of Claim 42 wherein said dielectric fluid is selected from the group consisting of tetrachloroethylene, carbon tetrachloride, pentane, 10 octane, decane, cyclohexane, benzene, xylene, sec-butylbenzene and Isopars.

          44.    The process of Claim 31, wherein said solvent is selected from the group consisting of ethanol, propanol, butanol, acetone, tetrahydrofuran, ketones, ethers, and esters.

15           45.    The process of Claim 34, wherein said first monomer is selected from the group consisting of styrene, methyl methacrylate, vinyl acetate, acrylate, ethyl vinylbenzene, vinylpyridine, and acrylonitrile, said crosslinker is selected from the group consisting of nonconjugated divinyl compounds, diacrylate 20 compounds, triacrylate compounds, dimethacrylate compounds and trimethacrylate compounds, said initiator is selected from the group consisting of 2,2'-azobisisobutyronitrile, 4,4'-azobis(r-cyanopentanic acid), 2,2'-azobis(2-

5 methylbutyronitrile), benzoyl peroxide, lauroyl peroxide and octanoyl peroxide, said stabilizer is selected from the group consisting of poly(vinyl pyrrolidone), homopolymers, copolymers, graft polymers, block polymers, poly(acrylic acid), poly(vinyl alcohol), poly(methacrylic acid) and sorbitan stearic acid monoester and said second monomer is acrylamide.

10 46. The process of Claim 45, wherein said functional monomer is selected from the group consisting of vinyl acetate, methyl methacrylate, acrylonitrile, acrylamide and dimethylaminopropylmethacrylamide and produce dielectric particles with a basic surface suitable for positive charging.

15 47. The process of Claim 36, wherein said first monomer is selected from the group consisting of styrene, methyl methacrylate, vinyl acetate, acrylate, ethyl vinylbenzene, vinylpyridine, and acrylonitrile, said crosslinker is selected from the group consisting of nonconjugated divinyl compounds, diacrylate compounds, triacrylate compounds, dimethacrylate compounds and trimethacrylate compounds, said initiator is selected from the group consisting of 2,2'-azobisisobutyronitrile, 4,4'-azobis(r-cyanopentanic acid), 2,2'-azobis(2-methylbutyronitrile), benzoyl peroxide, lauroyl peroxide and octanoyl peroxide, said stabilizer is selected from the group consisting of poly(vinyl pyrrolidone), homopolymers, copolymers, graft polymers, block polymers, poly(acrylic acid), poly(vinyl alcohol), poly(methacrylic acid) and sorbitan stearic acid monoester

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and said second monomer is methacrylic acid.

48. The process of Claim 47, wherein said functional monomer is selected from the group consisting of acrylic acid, sodium styrene sulfonate, maleic acid, chlorostyrene and vinyl alcohol, producing dielectric particles with an acidic surface suitable for negative charging.

## AMENDED CLAIMS

[received by the International Bureau on 20 July 1994 (20.07.1994);  
original claims 1-48 replaced by amended claims 1-42 (10 pages)]

1. A process for forming dielectric particles, comprising:  
admixing a first monomer and a crosslinker in a dispersion  
medium;  
5 adding an initiator and a stabilizer to said dispersion medium so  
that said first monomer polymerizes to form said dielectric particles; and  
introducing a second monomer and a functional monomer to said  
dispersion medium so that said second monomer and said functional monomer  
can at least partially polymerize and graft onto said dielectric particles to form  
10 a thin functional polymer coating which controls the charging of said dielectric  
particles.
2. The process of Claim 1, wherein said first monomer is a vinyl  
monomer.
- 15 3. The process of Claim 1, wherein said first monomer is selected  
from the group consisting of styrene, methyl methacrylate, vinyl acetate, acrylate,  
ethyl vinylbenzene, vinylpyridine, and acrylonitrile.
- 20 4. The process of Claim 3, wherein said crosslinker is a compound  
having a plurality of polymerizable double bonds.

5           5.     The process of Claim 4, wherein said crosslinker is selected from the group consisting of nonconjugated divinyl compounds, diacrylate compounds, triacrylate compounds, dimethacrylate compounds and trimethacrylate compounds.

10           6.     The process of Claim 5, wherein said initiator is selected from the group consisting of 2,2'-azobisisobutyronitrile, 4,4'-azobis(r-cyanopentanoic acid), 2,2'-azobis(2-methylbutyronitrile), benzoyl peroxide, lauroyl peroxide and octanoyl peroxide.

15           7.     The process of Claim 6, wherein said stabilizer is selected from the group consisting of poly(vinyl pyrrolidone), homopolymers, copolymers, graft polymers, block polymers, poly(acrylic acid), poly(vinyl alcohol), poly(methacrylic acid) and sorbitan stearic acid monoester.

          8.     The process of Claim 7, wherein said second monomer is acrylamide.

20           9.     The process of Claim 8, wherein said functional monomer is selected from the group consisting of vinyl acetate, methyl methacrylate, acrylonitrile, N-(iso-utoxymethyl) acrylamide and



dimethylaminopropylmethacrylamide and produce dielectric particles with a basic surface suitable for positive charging.

10. The process of Claim 9, wherein said particles produced are white.

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11. The process of Claim 10, further including the step of dying said white particles another color.

10

12. The process of Claim 11, wherein said dying is performed by immersing said white particles in an aqueous solution of a metal oxide.

13. The process of Claim 12, wherein said metal oxide is selected from the group comprising ruthenium tetroxide and osmium tetroxide.

15

14. The process of Claim 7, wherein said second monomer is methacrylic acid.

20

15. The process of Claim 14, wherein said functional monomer is selected from the group consisting of acrylic acid, sodium styrene sulfonate, maleic acid, chlorostyrene and vinyl alcohol, producing dielectric particles with an acidic surface suitable for negative charging.

16. The process of Claim 15, wherein said particles produced are white.

17. The process of Claim 16, further including the step of dying said white particles another color.

18. The process of Claim 17, wherein said dying is performed by immersing said white particles in an aqueous solution of a metal oxide.

19. The process of Claim 18, wherein said metal oxide is selected from the group comprising ruthenium tetroxide and osmium tetroxide.

20. The process of Claim 1, wherein said step of allowing said first monomer to polymerize includes agitating and warming said dispersion medium.

21. The process of Claim 1, further including the step of separating said dielectric particles from said dispersion medium by centrifuging and decanting the supernatant liquid.

22. A process for forming an electrophoretic fluid for use in an electrophoretic display, comprising:

admixing a first monomer and a crosslinker in a dispersion

medium; said dispersion medium being selected from the group consisting of alcohols, ethers, halogenated hydrocarbons, ketones and esters;

adding an initiator and a stabilizer to said dispersion medium so that said first monomer polymerizes to form dielectric polymer particles;

5 introducing a second monomer to said dispersion medium, said second monomer at least partially polymerizing and grafting upon said dielectric polymer particles;

separating said dielectric polymer particles from said dispersion medium; and

10 dispersing said dielectric particles in a dielectric fluid.

23. The process of Claim 22, further including the step of introducing a functional monomer to said dispersion medium when said second monomer is introduced.

15 24. The process of Claim 23, wherein said step of separating said dielectric particles from said dispersion medium includes centrifuging said dispersion medium and decanting the supernatant of said dispersion medium prior to said step of dispersing said particles in said dielectric fluid.

20 25. The process of Claim 24, further including the steps of resolating said dielectric particles after centrifuging said dispersion medium with a solvent

miscible with said dispersion medium and with said dielectric fluid and further centrifuging said particles and decanting said solvent prior to dispersing said particles in said dielectric fluid.

5                   26.    The process of Claim 25, further including the step of adding a charge control agent to said dielectric fluid after said step of dispersing in said dielectric fluid.

10                   27.    The process of Claim 26, wherein said charge control agent comprises an acidic surfactant and a positive charge control agent.

15                   28.    The process of Claim 27, wherein said positive charge control agent is selected from the group consisting of polyisobutylene succinic anhydride, cupric naphthenate, zirconium octoate, zinc octoate, calcium octoate, polyvinyl alcohol, polyacrylic acid, polymethacrylic acid and polyvinyl chloride.

                  29.    The process of Claim 26, wherein said charge control agent comprises a basic surfactant and a negative charge control agent.

20                   30.    The process of Claim 29, wherein said negative charge control agent is selected from the group consisting of polyisobutylene succinimide, barium petronate, barium sulfonate, barium

dinonylnaphthalensulfonate, metal oxides, polyvinyl pyridine, pyridine, lecithin, polyvinyl acetate, polyethelene oxide, polymethy methacrylate, polyacrylamide, polyesters and polyethers.

5           31.    The process of Claim 26, further including adding a dye to said dielectric fluid when said charge control agent is added.

10           32.    The process of Claim 31, wherein said dye is selected from the group consisting of Solvent Blue, Oil Blue A, Sudan Black B, and Sudan Red 7B.

          33.    The process of Claim 26, further including the step of adding a stabilizer to said dielectric fluid when said charge control agent is added.

15           34.    The process of Claim 33, wherein said stabilizer is selected from the group consisting of homopolymers, copolymers and block copolymers.

20           35.    The process of Claim 33, wherein said stabilizer is selected from the group consisting of poly(12-hydroxystearic acid)-graft-poly(methyl methacrylate-methacrylic acid), polystyrene-co-poly(vinyl pyridine) and poly(vinyl alcohol)-co-polyethylene.

36. The process of Claim 22, wherein said dielectric fluid is a non-polar solvent.

5           37. The process of Claim 36 wherein said dielectric fluid is selected from the group consisting of tetrachloroethylene, carbon tetrachloride, pentane, octane, decane, cyclohexane, benzene, xylene, sec-butylbenzene and Isopars.

10           38. The process of Claim 25, wherein said solvent is selected from the group consisting of ethanol, propanol, butanol, acetone, tetrahydrofuran, ketones, ethers, and esters.

15           39. The process of Claim 28, wherein said first monomer is selected from the group consisting of styrene, methyl methacrylate, vinyl acetate, acrylate, ethyl vinylbenzene, vinylpyridine, and acrylonitrile, said crosslinker is selected from the group consisting of nonconjugated divinyl compounds, diacrylate compounds, triacrylate compounds, dimethacrylate compounds and trimethacrylate compounds, said initiator is selected from the group consisting of 2,2'-azobisisobutyronitrile, 4,4'-azobis(r-cyanopentanic acid), 2,2'-azobis(2-methylbutyronitrile), benzoyl peroxide, lauroyl peroxide and octanoyl peroxide, said stabilizer is selected from the group consisting of poly(vinyl pyrrolidone), homopolymers, copolymers, graft polymers, block polymers, poly(acrylic acid),

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poly(vinyl alcohol), poly(methacrylic acid) and sorbitan stearic acid monoester and said second monomer is acrylamide.

5           40.    The process of Claim 39, wherein said functional monomer is selected from the group consisting of vinyl acetate, methyl methacrylate, acrylonitrile, acrylamide and dimethylaminopropylmethacrylamide and produce dielectric particles with a basic surface suitable for positive charging.

10           41.    The process of Claim 30, wherein said first monomer is selected from the group consisting of styrene, methyl methacrylate, vinyl acetate, acrylate, ethyl vinylbenzene, vinylpyridine, and acrylonitrile, said crosslinker is selected from the group consisting of nonconjugated divinyl compounds, diacrylate compounds, triacrylate compounds, dimethacrylate compounds and trimethacrylate compounds, said initiator is selected from the group consisting of  
15   2,2'-azobisisobutyronitrile, 4,4'-azobis(r-cyanopentanic acid), 2,2'-azobis(2-methylbutyronitrile), benzoyl peroxide, lauroyl peroxide and octanoyl peroxide, said stabilizer is selected from the group consisting of poly(vinyl pyrrolidone), homopolymers, copolymers, graft polymers, block polymers, poly(acrylic acid), poly(vinyl alcohol), poly(methacrylic acid) and sorbitan stearic acid monoester  
20   and said second monomer is methacrylic acid.

42. The process of Claim 41, wherein said functional monomer is selected from the group consisting of acrylic acid, sodium styrene sulfonate, maleic acid, chlorostyrene and vinyl alcohol, producing dielectric particles with an acidic surface suitable for negative charging.

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FIG. 1

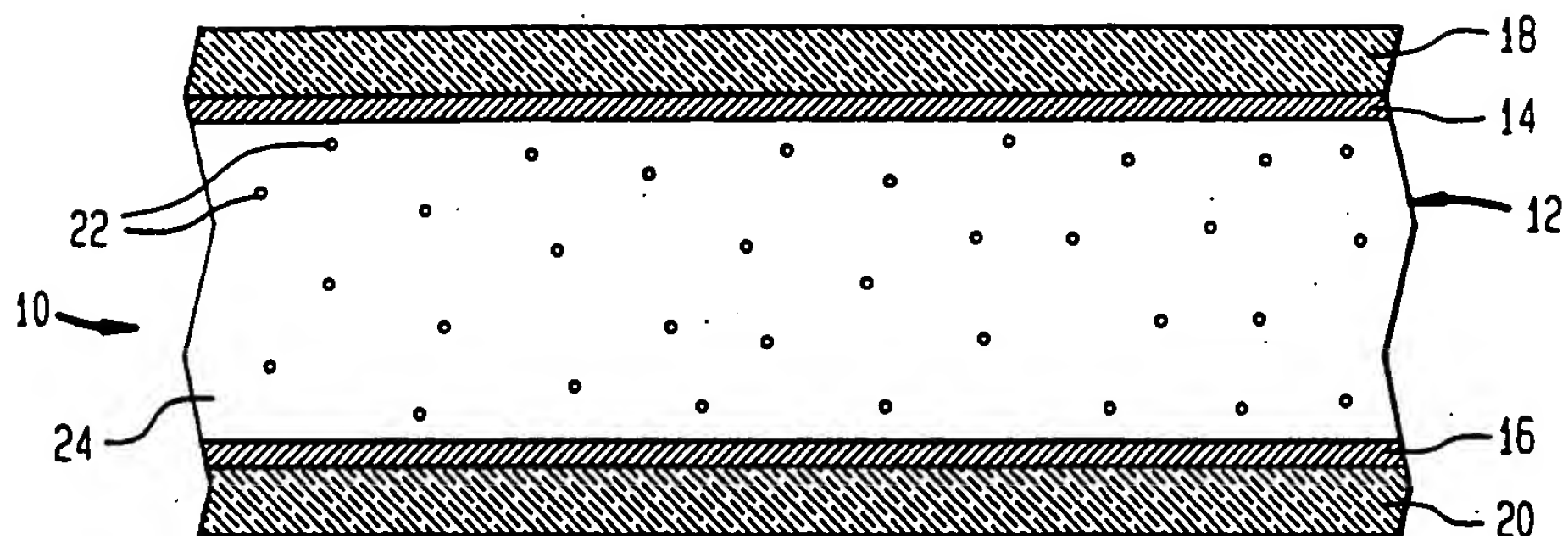
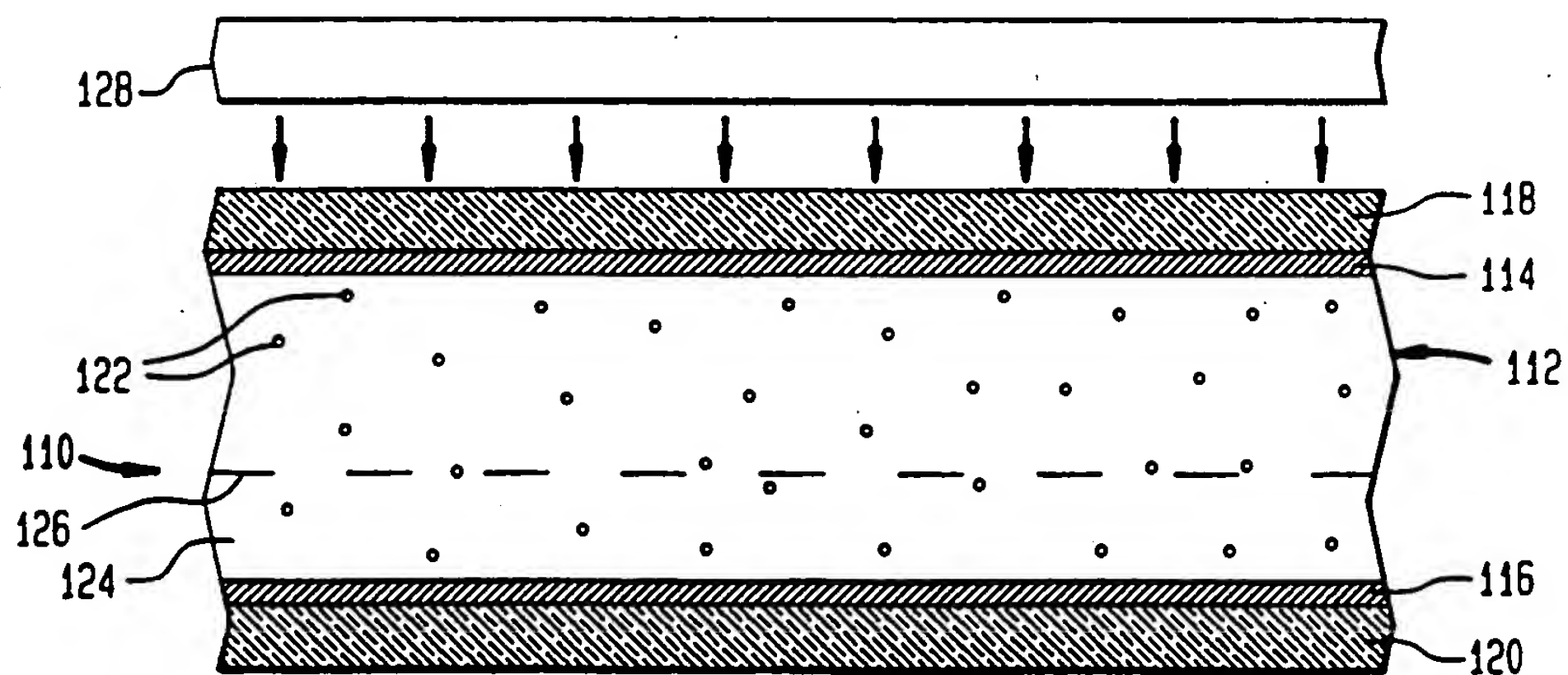


FIG. 2



## INTERNATIONAL SEARCH REPORT

International application No.  
PCT/US94/03813

## A. CLASSIFICATION OF SUBJECT MATTER

IPC(S) : C10M 169/04; C09K 3/00

US CL : 252/73, 77, 79, 572; 313/483; 204/299R

According to International Patent Classification (IPC) or to both national classification and IPC

## B. FIELDS SEARCHED

Minimum documentation searched (classification system followed by classification symbols)

U.S. : 252/73, 77, 79, 572; 313/483; 204/299R

Documentation searched other than minimum documentation to the extent that such documents are included in the fields searched  
NONE

Electronic data base consulted during the international search (name of data base and, where practicable, search terms used)

CAS styrene, methacrylic, crosslink, stabilize, electrophoretic

## C. DOCUMENTS CONSIDERED TO BE RELEVANT

Category*	Citation of document, with indication, where appropriate, of the relevant passages	Relevant to claim No.
Y	US, A, 4,992,192 (AHMED) 12 FEBRUARY 1991 (Note column 4, line 20 - column 5, line 19).	1-48
Y	Journal of Polymer Science: Part A: Polymer Chemistry Edition, Vol. 24, 1986, Tseng et al, "Uniform Polymer Particles by Dispersion Polymerization in Alcohol", pages 2995-3007.	1-27



Further documents are listed in the continuation of Box C.



See patent family annex.

* Special categories of cited documents	* T	later document published after the international filing date or priority date and not in conflict with the application but cited to understand the principle or theory underlying the invention
* A		
* E		
* L		
* F		
* P		
* X		document of particular relevance; the claimed invention cannot be considered novel or cannot be considered to involve an inventive step when the document is taken alone
	* Y	document of particular relevance; the claimed invention cannot be considered to involve an inventive step when the document is combined with one or more other such documents, such combination being obvious to a person skilled in the art
	* &	document member of the same patent family

Date of the actual completion of the international search

03 JUNE 1994

Date of mailing of the international search report

JUN 21 1994

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